POLBA MAHAVIDYALAYA Departmental Lesson Plan 2021-2022

Name of the Department : Department of Chemistry Name of the Programme : B.Sc.(General)

Name of the Course (Subject):CHEMISTRY...... Period of the Lesson Plan: July'21 to June'22

SEM-I			lectures	Teacher	Internal Assessment					
	GCC- 1A/	THEORY	64	Soumya Sinha Roy	24.12.21					
	- GE1	Organic Chemistry	32		-					
	<u> </u> 	1. Fundamentals of Organic Chemistry	04							
		2. Stereochemistry	04							
		3. Nucleophilic Substitution and Elimination Reactions	05							
				4. Aliphatic Hydrocarbons	02					
		Question-Answer Discussion	01							
		5. Alkanes	03		-					
								6. Alkenes	05	
		7. Alkynes	05		_					
		8. Some specific Reactions	03							
			1. Fundamentals of Organic Chemistry 2. Stereochemistry 3. Nucleophilic Substitution and Elimination Reactions 4. Aliphatic Hydrocarbons Question-Answer Discussion 5. Alkanes 6. Alkenes 7. Alkynes	1. Fundamentals of Organic Chemistry 04 2. Stereochemistry 04 3. Nucleophilic Substitution and Elimination Reactions 05 4. Aliphatic Hydrocarbons 02 Question-Answer Discussion 01 5. Alkanes 03 6. Alkenes 05 7. Alkynes 05	Organic Chemistry 1. Fundamentals of Organic Chemistry 2. Stereochemistry 3. Nucleophilic Substitution and Elimination Reactions 4. Aliphatic Hydrocarbons Question-Answer Discussion 5. Alkanes 6. Alkenes 7. Alkynes					

			Inorganic Chemistry	32		
			PRACTICAL	32 x 2 =64	Soumya Sinha Roy	
			Qualitative Analysis of Single Solid Organic Compound(s) [Known and Unknown Samples]	16×2 = 32		
			Inorganic Chemistry	16×2 = 32		
July'21 to Jan.'22	SEM-III	GCC-1C/ GE3	THEORY	64	Soumya Sinha Roy	16.12.21
			1. Aromatic Hydrocarbons	04		
			2. Organometallic Compounds	06		
			3. Aryl Halides	03		
			4. Alcohols, Phenols and Ethers:			
			(i) Alcohols	03		
			(ii) Phenols	03		
			(iii) Ethers	02		
			5. Carbonyl Compounds:			
			Aldehydes and Ketones (aliphatic and aromatic) :	02		
			(i) Preparations	03		
			(i) Preparations	03		

(ii) Reactions	03		
Thermodynamics upto 1st law	08		
	00		
Thermodynamics 2nd law	08		
Chemical Equilibrium	08		
Greimear Equinorram			
Ionic Equilibrium	08		
Question-Answer Discussion	03		
PRACTICAL	32 x 2 =64	Soumya Sinha Roy	
Identification of a pure organic compound (Known & Unknown Sample)	16		
Identification of a pure organic compound	16		
Measurement of pH of different solutions like aerated drinks, fruit juices, shampoos and soaps (use dilute solutions of soaps and shampoos to prevent damage to the glass electrode) using pH meter and compare it with the indicator method	04		
Practice Practice	04		
Preparation of buffer solutions and find the pH of an unknown buffer solution by colour matching method (Sodium acetate acetic acid)	04		
Practice	04		
Study of the solubility of benzoic acid in water	04		

			Practice	04		
			Preparation of buffer solutions and find the pH of an unknown buffer solution by colour matching method (Ammonium chloride ammonium hydroxide)	04		_
			Practice	04		
		SEC-1	Analytical Clinical Biochemistry	32	Soumya Sinha Roy	_
			Carbohydrates, Proteins, Structure of DNAto Gene Therapy, Enzymes	16		-
			Biochemistry of disease: A diagnostic approach by Blood/Urine analysis.	16		-
July'21 to S Jan.'22	SEM-V	DSE-1A	THEORY	64	Soumya Sinha Roy	02.12.21
			Inorganic Chemistry	32		-
			Transition Element	12		_
			Coordination Chemistry	12		
			Crystal Field Theory	08		
			Analytical Chemistry:	16		_
			Error Analysis	08]

Computer Application	08	
Industrial Chemistry	16	
Fuels	04	
Fertilizers	04	
Glass & Ceramics	04	
Cement	04	
PRACTICAL	32 x 2 =64	Soumya Sinha Roy
Titration of Na ₂ CO ₃ and NaHCO ₃ mixture vs HCl using phenolphthalein and methyl orange indicators.	10	
Practice	06	
Titration of HCl and CH ₃ COOH mixture vs NaOH using two different indicators to find the composition	10	
Practice	06	
Estimation of Total hardness of water sample by EDTA titration.	10	

		Practice	06		
		Estimation of available oxygen in pyrolusite.	10		
		Practice	06		
	SEC-3	Basic & Application of Computer in Chemistry i. Mathematics ii. Computer Programming	32 16 16	Soumya Sinha Roy	

POLBA MAHAVIDYALAYA Departmental Lesson Plan 2021-2022

Name of the Department : Department of Chemistry
Name of the Programme : B.Sc.(General)
Name of the Course (Subject) :CHEMISTRY......
Period of the Lesson Plan : July'21 to June'22

Academic Period	Class	Paper	Topic to be covered	No. of lectures	Name of the Teacher	Date of Internal Assessment
Feb'22 to SEM-II Jun.'22	SEM-II	GCC-1B/ GE2	THEORY	64	Soumya Sinha Roy	17.05.22
			Kinetic Theory of Gases and Real gases	08		
			Viscosity	03		
			Surface Tension	05		
			Chemical Bonding and Molecular Structure	16		

Chemical Kinetics	08	
Solid State	08	
Comparative study of p-block elements	16	
PRACTICAL	64	Soumya Sinha Roy
Determination of the surface tension of a liquid or a dilute solution using Stalagmometer.	04	
Study of the variation of surface tension of a detergent solution with concentration	04	
Determination of the relative and absolute viscosity of a liquid or dilute solution using an Ostwald's viscometer	04	
Study of the variation of viscosity of an aqueous solution with concentration of solute	04	
Study the kinetics of Iodide persulphate reaction	06	
Acid hydrolysis of methyl acetate with hydrochloric acid	04	
Compare the strengths of HCl and H2SO4 by studying kinetics of hydrolysis of methyl acetate	04	
Qualitative semi-micro analysis		
Basic Radicals: Na+, K+, Ca2+, Sr2+, Ba2+, Cr3+, Mn2+, Fe3+, Ni2+, Cu2+, NH4+.	16	
Acid Radicals: Cl-, Br-, I-, NO2-, NO3-, S2-, SO42-, PO43-, BO33-, H3BO3.	16	
Practice	02	

SEM-	IV GCC-1D/ GE4	THEORY	64	Soumya Sinha Roy	13.05.22
		Colligative Property	08		-
		Phase Equilibrium	08		-
		EMF	08		-
		Conductance	08		-
		Gravimetric Analysis	04		
		Chromatography	04		
		Volumetric Analysis	08		
		Environmental Chemistry: The Atmosphere	08		
		Environmental Chemistry: The Hydrosphere	08		
		PRACTICAL	64	Soumya Sinha Roy	
		Distribution Law	04		
		Practice	04		
		Determination of dissociation constant of a weak acid (Conductometrically)	04		
		Practice	04		
		Total hardness of water by EDTA titration	08		
		PH of an unknown solution by comparing color	08		
		potentiometric titration: Potassium dichromate vs. Mohr's salt	08		
		Practice	02		
		conductometric titration: Weak acid vs. strong base	06		
		Practice	02		
		Rate constant for the acid catalysed hydrolysis of an ester	08		
		Strength of the H2O2 sample	04		

		solubility of a sparingly soluble salt, e.g. KHTa	04		
	SEC-2	Drugs & Pharmaceuticals	32	Soumya Sinha Roy	
		Drug discovery, design and development; analgesics agents, antipyretic agents, anti- inflammatory agents	07		
		Antibiotics; antibacterial and antifungal agents; antiviral agents	06		
		Antiviral agents	03		
		Central Nervous System agents	03		
		Cardiovascular, etc	02		
		Antilaprosy	04		
		HIV-AIDS related drugs, etc.	04		
		Question-Answer Discussion	03		
SEM-VI	DSE-1B	THEORY	64	Soumya Sinha Roy	12.05.22
		1. Carboxylic Acids and Their Derivatives			
		a. Carboxylic acids (aliphatic and aromatic):	04		
		b. Carboxylic acid derivatives(aliphatic):	04		
		2. Amines and Diazonium Salts:			
		(a) Amines (aliphatic and aromatic);	03		
		(b) Diazonium salts	02		
		(c) Nitro compounds (aromatic)	03		

	3. Amino Acids	06	
	3.Amino Acids and Carbohydrates:		
	(ii) Carbohydrates	08	
	Polymers	06	
	Varnishes	02	
	Paints	04	
	Synthetic dyes	04	
	Drugs and pharmaceuticals	05	
	Food additives	03	
	Fats and oils	02	
	Soaps and detergents	03	
	Pesticides	03	
	Question-Answer Discussion	02	
	PRACTICAL	64	Soumya Sinha Roy
	Organic Chemistry(Practical)	16	
	Functional Group Organic Chemistry	16	
	Estimation of saponification value of oil/fat.	12	
	Practice	04	
	Estimation of acetic acid in commercial vinegar.	12	
	Practice	04	
SEC-4	Polymer Chemistry	32	Soumya Sinha Roy
	Introduction and history of polymeric materials	08	
	Functionality and its importance	08	

	Kinetics of Polymerisation	06	
	Determination of molecular weights	06	
	Properties of Polymers	04	

POLBA MAHAVIDYALAYA

Implementation of Departmental Lesson Plan 2021-2022

Name of the Department : Department of Chemistry

Name of the Programme: B.Sc.(General)

Name of the Course (Subject):CHEMISTRY.....

Period of the Lesson Plan: July'21 to June'22

Academic Period	Class	Paper	Topic covered	Topic Not covered	Reason for Not covered	Date of Internal Assessment	Remarks
July'21 to Jan.'22	SEM-I	GCC-1A/ GE1	THEORY			24.12.21	
			Organic Chemistry	All completed			
			1. Fundamentals of Organic Chemistry				
			2. Stereochemistry				
			3. Nucleophilic Substitution and Elimination Reactions				
			4. Aliphatic Hydrocarbons				
			Question-Answer Discussion				
			5. Alkanes				
			6. Alkenes				
			7. Alkynes				

			8. Some specific Reactions			
			1. Fundamentals of Organic Chemistry			
			2. Stereochemistry			
			3. Nucleophilic Substitution and Elimination Reactions			
			4. Aliphatic Hydrocarbons			
			Question-Answer Discussion			
			Inorganic Chemistry	All completed		
			PRACTICAL			
			Qualitative Analysis of Single Solid Organic Compound(s) [Known and Unknown Samples]	All completed		
			Inorganic Chemistry	All completed		
5	SEM-III	GCC-1C/ GE3	THEORY		16.12.21	
			1. Aromatic Hydrocarbons	All completed		
			2. Organometallic Compounds	All completed		
			3. Aryl Halides	All completed		
			4. Alcohols, Phenols and Ethers:	All completed		

Question-Answer Discussion	All completed	
Question miswer Discussion		
PRACTICAL		
	All completed	
compound (Known &		
Unknown Sample)	A11 1 1 1	
	All completed	
compound		
Measurement of pH of	All completed	_
different solutions like aerated	The completed	
drinks, fruit juices, shampoos		
and soaps (use dilute solutions		
of soaps		
and shampoos to prevent		
damage to the glass electrode)		
using pH meter and compare it		
with the indicator method		
Practice		
Preparation of buffer solutions	All completed	-
and find the pH of an unknown	7 in completed	
buffer solution by colour		
matching method (Sodium		
acetate acetic acid)		
Practice		
	All completed	
benzoic acid in water		

			Practice				
			Preparation of buffer solutions and find the pH of an unknown buffer solution by colour matching method (Ammonium chloride ammonium hydroxide) Practice	All completed			
		GEG1					
		SEC1		Analytical Clinical Biochemistry Carbohydrates, Proteins, Structure of	No student		
				DNAto Gene Therapy, Enzymes Biochemistry of disease: A diagnostic			
				approach by Blood/Urine analysis.			
July'21 to Jan.'22	SEM-V	DSE-1A	THEORY			02.12.21	
			Inorganic Chemistry	All completed			
			Transition Element				
			Coordination Chemistry				
			Crystal Field Theory				
			Analytical Chemistry:	All completed			
			Error Analysis				

Computer Application		
Industrial Chemistry	All completed	
Fuels		
Fertilizers		
Glass & Ceramics		
Cement		
PRACTICAL		
Titration of Na ₂ CO ₃ and NaHCO ₃ mixture vs HCl using phenolphthalein and methyl orange indicators. Practice	All completed	
Titration of HCl and CH ₃ COOH mixture vs NaOH using two different indicators to find the composition	All completed	
Practice		
Estimation of Total hardness of water sample by EDTA titration.	All completed	
Practice		

		Estimation of available oxygen in pyrolusite.	All completed		
		Practice			
	SEC-3		Basic & Application of Computer in Chemistry	No student	
			Mathematics		
			Computer Programming		

POLBA MAHAVIDYALAYA Implementation of Departmental Lesson Plan 2021-2022

Name of the Department: Department of Chemistry

Name of the Programme : B.Sc.(General)

Name of the Course (Subject):CHEMISTRY.....

Period of the Lesson Plan: July'21 to June'22

Academic Period	Class	Paper	Topic covered	Topic Not covered	Reason for Not covered	Date of Internal Assessment	Remarks
Feb'22 to Jun.'22	SEM-II	GCC- 1B/ GE2	THEORY			17.05.22	
			Kinetic Theory of Gases and Real gases	All completed			
			Viscosity	All completed			

Surface Tension	All completed	
Surface Telision	The completed	
	411	_
Chemical Bonding and	All completed	
Molecular Structure		
Chemical Kinetics	All completed	
Solid State	All completed	_
Sona state		
	All completed	_
Comparative study of p-block	All completed	
elements		
PRACTICAL		
Determination of the surface	All completed	7
tension of a liquid or a dilute		
solution using Stalagmometer.		
Cturder of the acceptations of	All completed	-
Study of the variation of	All completed	
surface tension of a detergent solution with		
concentration		
Concenti ation		

Determination of the relative and absolute viscosity of a liquid or dilute solution using an Ostwald's viscometer	All completed	
Study of the variation of viscosity of an aqueous solution with concentration of solute	All completed	
Study the kinetics of Iodide persulphate reaction	All completed	
Acid hydrolysis of methyl acetate with hydrochloric acid	All completed	
Compare the strengths of HCl and H2SO4 by studying kinetics of hydrolysis of methyl acetate	All completed	
Qualitative semi-micro analysis	All completed	
Basic Radicals: Na+, K+, Ca2+, Sr2+, Ba2+, Cr3+, Mn2+, Fe3+, Ni2+, Cu2+, NH4+.	All completed	
Acid Radicals: Cl-, Br-, I-, NO2-, NO3-, S2-, SO42-, PO43-, BO33-, H3BO3.	All completed	

		Practice		
SEM-IV	GCC-	THEORY		13.05.22
JLIVI-I V	1D/ GE4	meoki		13.03.22
		Colligative Property	All completed	
		Phase Equilibrium	All completed	
		EMF	All completed	
		Conductance	All completed	
		Gravimetric Analysis	All completed	
		Chromatography	All completed	
		Volumetric Analysis	All completed	
		Environmental Chemistry: The Atmosphere	All completed	
		Environmental Chemistry: The Hydrosphere	All completed	
		PRACTICAL		
		Distribution Law	All completed	
		Practice		
		Determination of dissociation constant of a weak acid (Conductometrically)	All completed	
		Practice		
		Total hardness of water by EDTA titration	All completed	
		PH of an unknown solution by comparing color	All completed	

	potentiometric titration: Potassium dichromate vs. Mohr's salt Practice	All completed		
	conductometric titration: Weak acid vs. strong base	All completed		
	Practice Rate constant for the acid	All completed		
	catalysed hydrolysis of an ester Strength of the H2O2 sample	All completed		
	solubility of a sparingly	All completed		
SEC-2	soluble salt, e.g. KHTa	Drugs & Pharmaceuticals	No student	
		Drug discovery, design and development; analgesics agents, antipyretic agents, anti- inflammatory agents	No student	
		Antibiotics; antibacterial and antifungal agents; antiviral agents	No student	
		Antiviral agents	No student	
		Central Nervous System agents	No student	
		Cardiovascular, etc	No student	
		Antilaprosy	No student	

		HIV-AIDS related drugs, etc.	No student	
		Question-Answer Discussion	No student	
M-VI DSE-	1B THEORY			12.05.22
	1. Carboxylic Acids and Their Derivatives	All completed		
	a. Carboxylic acids (aliphatic and aromatic):	All completed		
	b. Carboxylic acid derivatives(aliphatic):	All completed		
	2. Amines and Diazonium Salts:	All completed		
	(a) Amines (aliphatic and aromatic);	All completed		
	(b) Diazonium salts	All completed		
	(c) Nitro compounds (aromatic)	All completed		
	3. Amino Acids	All completed		
	3.Amino Acids and Carbohydrates:	All completed		
	(ii) Carbohydrates	All completed		
	Polymers	All completed		
	Varnishes	All completed		
	M-VI DSE-	1. Carboxylic Acids and Their Derivatives a. Carboxylic acids (aliphatic and aromatic): b. Carboxylic acid derivatives(aliphatic): 2. Amines and Diazonium Salts: (a) Amines (aliphatic and aromatic); (b) Diazonium salts (c) Nitro compounds (aromatic) 3. Amino Acids 3. Amino Acids and Carbohydrates: (ii) Carbohydrates	M-VI DSE-1B THEORY 1. Carboxylic Acids and Their Derivatives a. Carboxylic acids (aliphatic and aromatic): b. Carboxylic acid derivatives(aliphatic): 2. Amines and Diazonium Salts: (a) Amines (aliphatic and aromatic); (b) Diazonium salts All completed (c) Nitro compounds (aromatic) 3. Amino Acids All completed 3. Amino Acids and Carbohydrates: (ii) Carbohydrates All completed All completed All completed	M-VI DSE-1B THEORY 1. Carboxylic Acids and Their Derivatives a. Carboxylic acids (aliphatic and aromatic): b. Carboxylic acid derivatives(aliphatic): 2. Amines and Diazonium Salts: (a) Amines (aliphatic and aromatic); (b) Diazonium salts All completed (c) Nitro compounds (aromatic) 3. Amino Acids All completed 3. Amino Acids and Carbohydrates: (ii) Carbohydrates All completed All completed All completed

	Paints	All completed		
	Synthetic dyes	All completed		
	Drugs and pharmaceuticals	All completed		
	Food additives	All completed		
	Fats and oils	All completed		
	Soaps and detergents	All completed		
	Pesticides	All completed		
	Question-Answer Discussion	All completed		
	PRACTICAL			
	Organic Chemistry(Practical)	All completed		
	Functional Group Organic Chemistry	All completed		
	Estimation of saponification value of oil/fat.	All completed		
	Practice			
	Estimation of acetic acid in commercial vinegar.	All completed		
	Practice			

POLBA MAHAVIDYALAYA Implementation of Departmental Lesson Plan 2021-2022

Name of the Department: Department of Chemistry

Name of the Programme : B.Sc.(General)

Name of the Course (Subject):CHEMISTRY.....

Period of the Lesson Plan: July'21 to June'22

Academic Period	Class	Paper	Topic covered	Topic Not covered	Reason for Not covered	Date of Internal Assessment	Remarks
Feb'22 to Jun.'22	SEM-VI	SEC-4		Polymer Chemistry	No student		
				Introduction and history of polymeric materials	No student		
				Functionality and its importance	No student		
				Kinetics of Polymerisation	No student		
				Determination of molecular weights	No student		
				Properties of Polymers	No student		